

April 5th, 2000

California Pacific Research, Inc.
300 Brinkby, Suite 200
Reno, NV 89509

ATTENTION: ROBERT MURPHY-PRESIDENT/CEO.

**SUBJECT: SUMMARY OF 2 TESTS ON YOUR CORROSION
PROTECTION COMPOUND.**

Dear Mr. Murphy,

Reference the above and the sample sent to me for testing the corrosion protection of the above compound as compared to Rustoleum.

Summary of Expt. #1:

Two stainless steel rods were sprayed with Rustoleum and coated with your compound, dried for 12 hours at 20 degrees C, and then immersed in hydrochloric acid, approximately 5% by weight.

After immersing the same for 8.5 hours, at room temperature, the Rustoleum coating peeled off, whereas your compound's coating remained intact.

**CONCLUSION: YOUR COMPOUND WORKS BETTER THAN
RUSTOLEUM.**

Summary of Expt. #2:

Three identical six long black iron nipples were sanded, washed and dried, then immersed in about 10% hydrochloric acid.

The first nipple was sanded and immersed in the acid as-is, whereas the second nipple was coated, twice, with Rustoleum and

Expt. #2 continued:

The third nipple was coated with your product.

The nipples were dried, two with the coatings, for two hours and 26 minutes at 30 degrees C, thereafter the nipples were immersed in the above acid, SG of 1.052.

After four hours, the Rustoleum was peeling, whereas your compound was intact.

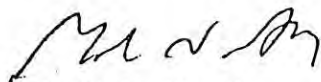
After 6 hours and 20 minutes, Rustoleum was completely removed, whereas the as-is nipple showed normal corrosion and your product's coating was intact.

After 16 hours and 27 minutes your coating was still intact, whereas, the Rustoleum was long gone, as above, and the as-is showed accelerated acid corrosion.

CONCLUSION: YOUR COMPOUND WORKS BETTER THAN RUSTOLEUM.

A formal report with photographs will be sent to you this week.

Sincerely,



Mobashir N. Ahmad
Proprietor
Chemicals & Metals, Inc.
Environmental and Metallurgical Consultant
C/O
University of Nevada at Reno
MS 199
Reno, Nevada 89557
775 853-8999